Acids Naming Practice and Review Name: \_\_\_\_\_\_\_\_\_\_

|  |  |  |  |
| --- | --- | --- | --- |
| Example | Anion Ending |  | Acid Name |
| HCl | -ide (binary) |  | Hydro\_\_\_\_\_ic acid |
| HClO3 | -ate (polyatomic) |  | \_\_\_\_\_\_ic acid |
| HClO2 | -ite (polyatomic) |  | \_\_\_\_\_\_ous acid |

Practice: Write the correct name or formula.

H2SO3 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

H2SO4 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

H2S \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

H3P\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

H3PO4 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

H3PO3 \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

HC2H3O2 (CH3COOH) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

hydrofluoric acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_

hydroselenic acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_

carbonic acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

carbonic acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

nitrous acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

nitric acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

hydrobromic acid \_\_\_\_\_\_\_\_\_\_\_\_\_\_